

INVESTIGATION OF I-V CHARACTERISTICS OF CARBON NANOTUBES AND TIN OXIDE NANOWIRES HETEROJUNCTION FOR GAS SENSING APPLICATIONS

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Abstract. In this study, heterojunctions of carbon nanotubes (CNTs) and tin oxide nanowires (SnO₂) were investigated to determine their electrical performance for gas sensing applications. The I–V characteristics of the heterojunction were measured in air and H₂S gas. The electrical parameters such as barrier height ϕ_B , ideality factor n , series resistance R_s , and reverse saturation current I_o were extracted from the I-V plots. In H₂S gas, the barrier height ϕ_B and series resistance R_s are lower compared to air, whereas the ideality factor n and saturation current I_o are higher. These findings are essential for promoting technologies aimed at forming high-quality heterojunctions for gas sensing applications.

Keywords: Heterojunction, carbon nanotubes, SnO₂ nanowires, gas sensing.

1. Introduction

Heterojunctions are the technological keys to many electronic and optoelectronic devices. In recent decades, low-dimensional nanostructures of many materials have been successfully fabricated. Nano heterojunctions have also been studied to improve the gas sensing characteristics [1]. The potential barrier height of these heterojunctions is very sensitive to external agents. Especially at the nanoscale, where the material is very porous, gases are easily absorbed and drastically change the potential barrier height of the junction. This makes the sensor more sensitive [2-4]. Following this trend, heterojunctions of individual materials such as semiconductor metal oxide nanowires (SMO-NWs) [5], CNTs [6], Graphene (GP) [7], transition metal Dichalcogenides (TMDs) [8] were investigated for application in new sensor generations. Due to many

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interesting properties, the heterojunctions of metal oxides semiconductor nanowires and CNTs are attracting interest in applied research in a wide range of electronic components [9], including diodes or supercapacitors on flexible substrates [10-14], electrochemical devices [15], and field emission [16]. Carbon nanomaterials can exhibit p-type semiconductor or metallic properties depending on the structure [17] so the junctions between carbon nanomaterials and metal-semiconductor oxides can be p-n or Schottky. Such heterojunctions are not only fundamentally attractive from the perspective of electronic devices but also very promising for sensing applications. Research by Dai et al. showed that the heterojunction between nano-bar α -Fe₂O₃ and CNTs was highly sensitive to acetone gas and the detection limit reached 500 ppb concentration [18]. Another study by Lupan et al. [19] showed that transitions between ZnO nanowires and CNTs could detect NH₃ gas at room temperature with a detection limit of 400 ppb. Research by Li et al. [20] showed the ability of NO₂ gas sensitivity at room temperature of the sensor based on the transition between ZnO nanowires and m-SWCNTs metal electrodes. The response of the ZnO/m-SWCNTs structure to 2.5 ppm NO₂ at room temperature reached about 52%. Recently, we successfully fabricated the SnO₂/CNTs heterojunctions for NO₂ gas sensing applications [21-23], the results show that these heterojunctions have potential applications in highly sensitive gas sensors. However, the gas sensing mechanisms of these structures have not been clarified. Therefore, understanding the fundamental physical and electrical properties of the SnO₂/CNTs heterojunction is important for the development of applications based on this structure.

To explain the sensing mechanism of the gas sensors based on heterojunction structures, most studies have suggested that the change in barrier height at the contact between the two materials plays an important role in the sensitivity of the gas [24, 25]. In this study, the I-V characteristics of CNTs/SnO₂ were investigated in different gas environments to determine the heterojunction characteristics such as saturation leakage current, ideal factor, barrier height, and series resistance to elucidate the gas sensing mechanism.

2. Contents

2.1. Experiments

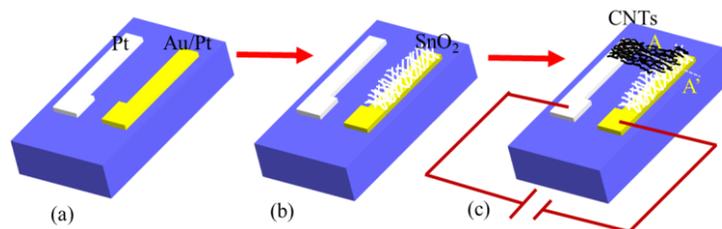


Figure 1. Scheme for fabrication of the heterojunction: (a) Pt and Au-catalyzed Pt electrodes, (b) selective growth of SnO₂ nanowires, and (c) CNTs/SnO₂ heterojunction

The heterojunctions CNTs/SnO₂ are fabricated on electrodes with the structure as shown in Figure 1. First, the sensor chip with one Pt electrode and one Au-catalyzed Pt electrode (Figure 1(a)) is fabricated by conventional photolithography and sputtering methods. SnO₂ nanowires were fabricated directly on an Au-catalyzed Pt electrode (Figure 1(b)) by

the thermal vapor deposition method [26]. The gap between the two electrodes is designed to be wide enough and the technological parameters in the fabrication process are controlled such that the SnO₂ nanowires are dense enough to completely cover the electrode but not long enough to bridge the two electrodes [22]. In detail, high purity tin powders (99.9%, Sigma Aldrich) were used as the reaction source. For growing nanowires, 0.1 g of tin powders were loaded in an alumina boat, which was then placed inside the center of a horizontal quartz tube furnace. The distance between the electrodes and the alumina boat was about 2 cm. The whole system was evacuated to a pressure of 1.5×10^{-1} Torr using a rotary pump and then purged with high purity argon. Then the furnace was heated up to 750 °C for 20 mins from room temperature and maintained for 15 mins under a flow of oxygen at 0.5 sccm. Finally, the system was cooled down naturally to room temperature. In this study, the commercial CNTs with diameters of about 60-100 nm were dispersed in an aqueous solvent and surfactant P123. After fabrication of SnO₂ nanowires, CNTs were coated onto the electrode by dip-coating to form CNTs/SnO₂ heterojunctions (Figure 1(c)). The complete device was annealed at 350 °C for 2 hours to remove solvents.

The characteristics of the CNTs/SnO₂ heterojunctions have been investigated by FE-SEM methods (JEOL 7600F) and Raman (Micro-Raman InVia, RENISHAW, H44840, Laser 633 nm). The I-V characteristics of the junctions were investigated at different temperatures, voltages, and gas conditions using a Keithley 2602 source and controlled by the Labview program. To measure the I-V characteristic, the CNTs/SnO₂ heterojunction is biased by applying DC voltage to the two electrodes which are shown in Figure 1c. The junction is forward biased by connecting CNTs to the positive terminal, the SnO₂ nanowires to the negative terminal of the power supply, and vice versa in the case of reverse bias.

2.2. Results and discussion

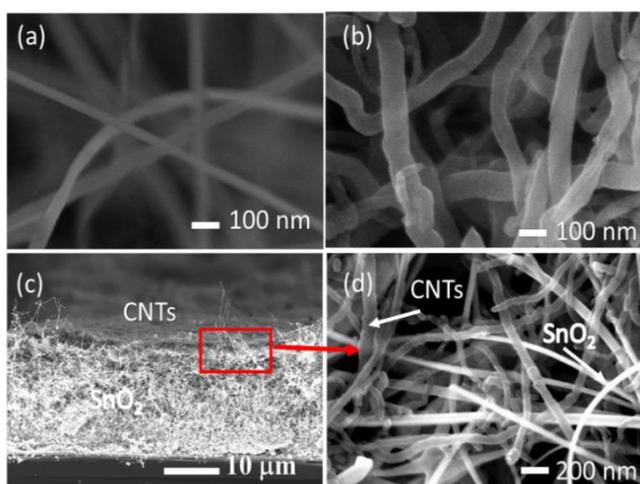


Figure 2. (a) SnO₂ nanowires grown directly on a Pt electrode, (b) The commercial CNTs diameter $d:60-100$ nm. (c) The cross-section SEM image of the CNTs/SnO₂ heterojunctions, (d) enlarged image of the CNTs randomly networked on top of SnO₂ nanowires

The morphology of SnO₂ nanowires, the commercial CNTs, and CNTs/SnO₂ heterojunction were observed by SEM and the results are presented in Figure 2. SEM image of SnO₂ directly grown on Pt electrode by the thermal vapor deposition method in Figure 2 (a) shows that the diameter of the nanowires is about 40-60 nm. The high-magnification SEM image in Figure 2 (b) revealed that CNTs with a diameter of 60-100 nm are uniformly dispersed and have high porosity. The SEM image of A-A' cross-section (Figure 1(c)) of the heterojunction was presented in Figure 2c, which can be seen in the high density SnO₂ nanowires completely covering the Pt electrode and CNTs form a film on top of SnO₂. The enlarged image in Figure 3 (d) shows that the CNTs randomly networked come into contact with on-chip grown SnO₂ nanowires at the junction area.

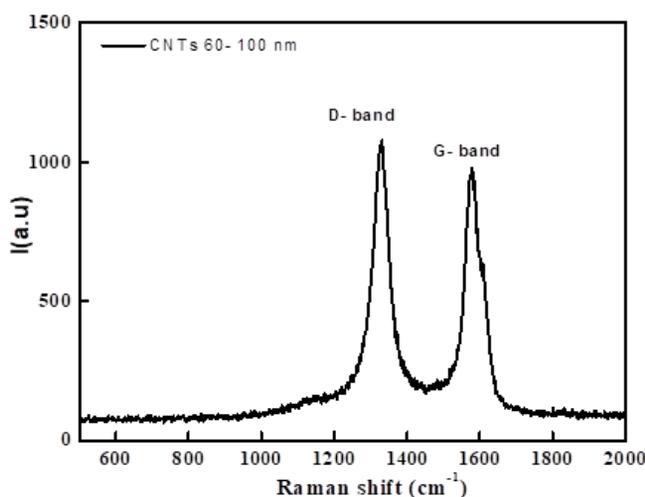


Figure 3. Raman spectra were taken on SnO₂/CNTs heterojunctions

The Raman spectra of the heterojunctions are presented in Figure 3. The spectrum reveals two main typical modes of MWCNTs: the D band at 1342 cm⁻¹, which is activated by the presence of disorder in carbon systems, and the G band at 1580 cm⁻¹, which is assigned to the in-plane vibration of the C-C.

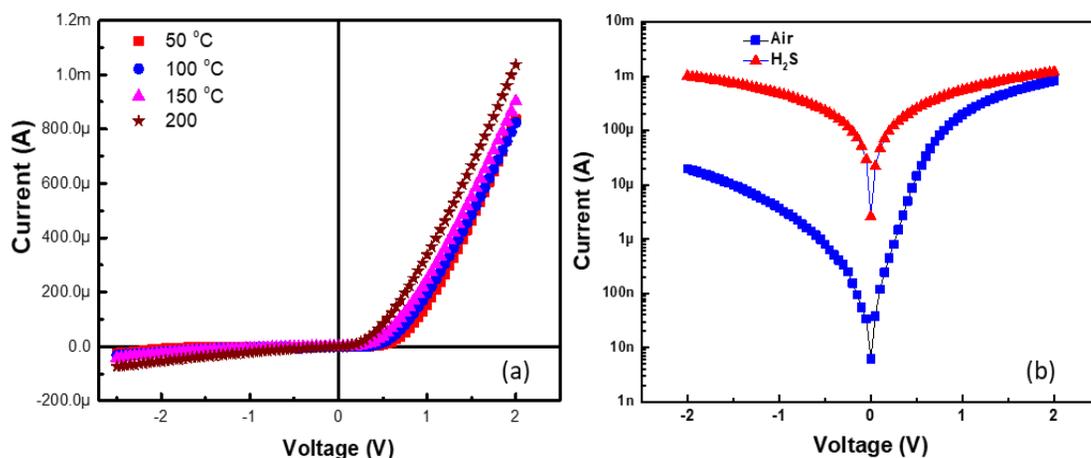


Figure 4. (a) I-V plot of CNTs/SnO₂ heterojunction in the air at different temperatures, (b) I-V plots of CNTs/SnO₂ heterojunction in air and 0.25 ppm H₂S at 100 °C

The I-V characteristics of the CNTs/SnO₂ heterojunctions in the air at the temperatures of 50 °C, 100 °C, 150 °C, and 200 °C are presented in Figure 4 (a). It can be seen that, in the bias voltage range from -2 V to +2 V, the CNTs/SnO₂ heterojunctions exhibit good rectification at operating temperatures from 50 °C to 200 °C. Since the contacts between SnO₂ and CNTs with the Pt electrode are Ohmic [21], the rectification property of the structure is due to the formation of a potential barrier at the junction between the SnO₂ nanowires and CNTs. It also proves that the CNTs network is only in contact with the left Pt electrode and in contact with the SnO₂ nanowires that have grown directly on the right Pt electrode (as shown in Figure 1 (c)). The I-V characteristics of the CNTs/SnO₂ heterojunction were studied at 100 °C in an H₂S gas concentration of 0.25 ppm. The results in Figure 4 (b) show that in H₂S gas, the current through the junction is increased compared to in the air. The variation of leakage current in analytical gas is larger than in forward-biased current. At a bias voltage of -2 V, the leakage current in the air has a value of about 20 μA in H₂S 0.25 ppm the leakage current increased to 1 mA.

The CNTs/SnO₂ heterojunction can be described using the equivalent circuit model shown in Figure 5. The diode component in the circuit represents the interface between CNTs and SnO₂. R_{CNTs} and R_{SnO2} are series resistance of CNTs and SnO₂. From the I-V characteristics, the diode parameters of the equivalent circuit model including saturation leakage current, ideality factor, potential barrier height, and series resistance of the junction can be calculated.

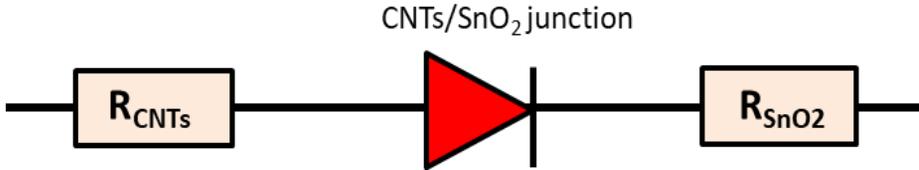


Figure 5. Equivalent circuit model of the CNTs/SnO₂ heterojunctions

According to many studies, the relationship between the current and the voltage of the nano-heterojunction can be described according to the thermal emission theory (TE) model [3, 27, 28]. For bias voltage $V > 3k_B T/q$, the relation between the applied forward bias and current can be calculated by the following equation:

$$I = I_o \exp\left(\frac{qV}{nk_B T}\right) \quad (1)$$

where

$$I_o = AA^* T^2 \exp\left(\frac{-q\phi_B}{k_B T}\right) \quad (2)$$

where I_o (A) is the saturation current, n is the ideality factor, k_B (J/K) is Boltzmann constant, q (C) is the electron charge, T (K) is absolute temperature, A (cm²) is the area of the junction estimated by the area of the electrode, A^* (Acm⁻²K⁻²) is effective Richardson constant, and ϕ_B (eV) is the barrier height of the heterojunction.

Taking the natural logarithm of (1) we have

$$\ln I = \ln I_o + \frac{qV}{nk_B T} \quad (3)$$

The linear fitting on the $\ln I$ vs. V plots of the heterojunction in the linear region of experimental data in air and 0.25 pp H_2S gas at a temperature of 100 °C was presented in Figure 6. We determined the values of n and I_o from the slope and the y-axis intercept of the fitted straight line.

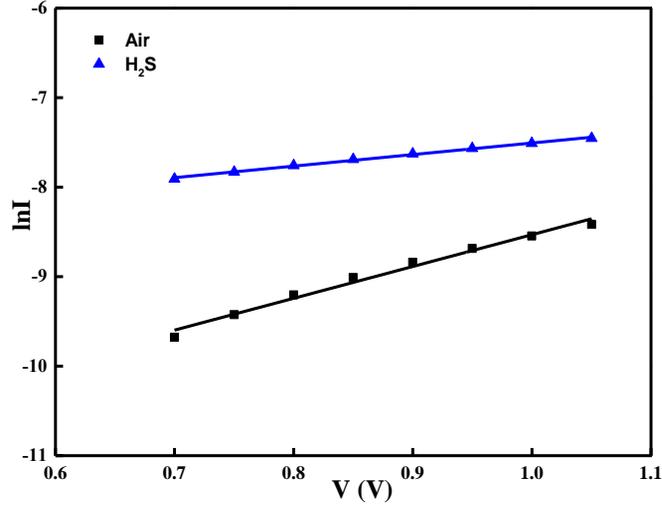


Figure 6. $\ln I$ - V plot of the CNTs/ SnO_2 heterojunction in Air and 0.25 ppm H_2S at a temperature of 100 °C

If we consider the effect of series resistance R_s caused by CNTs and SnO_2 , equation (1) becomes

$$I = I_o \exp\left(\frac{q(V - IR_s)}{nk_B T}\right) \quad (4)$$

The forward bias current-voltage characteristics due to thermionic emission of the heterojunction with series resistance can be expressed as Cheung's functions [29, 30].

$$\frac{dV}{d(\ln I)} = IR_s + \frac{nk_B T}{q} \quad (5)$$

$$H(I) = V - \left(\frac{nk_B T}{q}\right) \ln\left(\frac{I}{AA^* T^2}\right) \quad (6)$$

$$H(I) = IR_s + n\phi_B \quad (7)$$

In Figure 7(a-b), the intercept and slope of the line can be determined by linearly fitting the graph of equation (5) $dV/d(\ln I)$ vs. I . $H(I)$ is defined from (6) using the value of n and the data of I - V characteristics. The slope and y-axis intercept of the linear fitting of Plotting $H(I)$ vs. I should give the heterojunction barrier height.

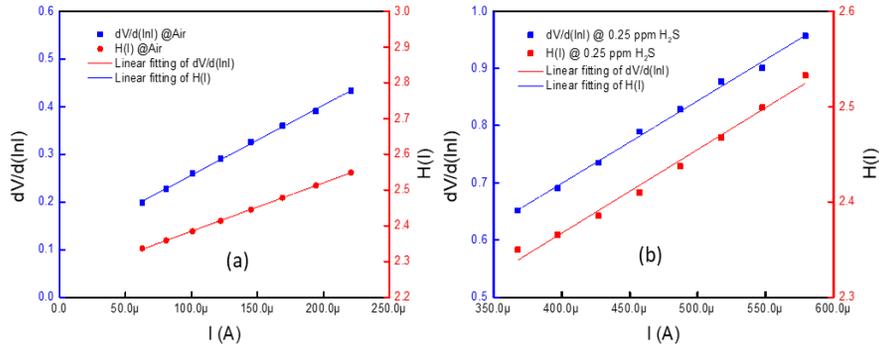


Figure 7. The $dV/d(\ln I)$ vs. I and $H(I)$ vs. I plots of the CNTs/SnO₂ heterojunction in the air (a) and 0.25 ppm H₂S at a temperature of 100 °C

The parameters of the CNTs/SnO₂ heterojunction were calculated based on the I-V characteristics and are presented in Table 1. It was observed that values of I_o , n , R_s , and ϕ_B are all changed in the analytical gas environment. In H₂S gas, the barrier height ϕ_B and series resistance R_s are decreased compared to in air while ideality factor n and saturation current I_o increased. The barrier height is observed to decrease from 0.66 eV in the air to 0.53 eV in 0.25 ppm H₂S. The saturation current I_o increased from $5.65 \cdot 10^{-6}$ A in the air to $1.51 \cdot 10^{-4}$ A in 0.25 ppm H₂S. Alternatively, a rise in the ideality factor in H₂S gas suggests a reduction in the rectification behavior of the heterojunction. In addition, the series resistance does not change appreciably, which shows that the gas sensitivity mechanism is dictated by the contact between CNTs and SnO₂ nanowires.

Table 1. Parameters obtained from I-V characteristics of CNTs/SnO₂ heterojunction in Air and 0.25 ppm H₂S

Parameters	Air	0.25 ppm H ₂ S
I_o (A)	$5.65 \cdot 10^{-6}$	$1.51 \cdot 10^{-4}$
n	3.4	3.8
R_s (Ω)	1466	878
ϕ_B (eV)	0.66	0.53

3. Conclusions

In summary, the CNTs/SnO₂ heterojunctions have been successfully fabricated which exhibit good rectification at operating temperatures from 50 °C to 200 °C. An equivalent circuit model of the junction was proposed and the diode parameters including the ideality factor, the barrier height, and the series resistance were extracted based on the thermionic emission theory from I-V measurements in air and H₂S gas. As a result, the barrier height decreased from 0.66 eV in the air to 0.53 eV in 0.25 ppm H₂S. The saturation current I_o increased from $5.65 \cdot 10^{-6}$ A in the air to $1.51 \cdot 10^{-4}$ A in 0.25 ppm H₂S. The ideality factor in H₂S is higher than in the air. However, the series resistance in H₂S does not change significantly compared to in the air. These results are of great importance for assisting the development of technologies aimed at forming high-quality heterojunctions for gas sensing applications.

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